

A Versatile High Yield Photochemical Synthesis of 5-Substituted 6-Azaauracils.

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Received March 18, 1974

Sir:

There has been extensive interest in *as*-triazines (often referred to as 6-azapyrimidines) as antiviral agents and as antimetabolites of pyrimidine bases (2). In spite of the biological interest in these systems, most syntheses of substituted 6-azauracils (*as*-triazine-3,5-(2*H*,4*H*)diones) are limited to ring closure processes of α -ketoacid derivatives (3); synthetic routes to functionalized 5-substituted 6-azauracils are uncommon. In connection with photochemical studies of azapyrimidine systems (4), we have noted a mild, facile conversion of the parent 6-azauracil to functionalized 5-substituted derivatives which we wish to report here. Appropriate reactions at the 5-substituent should provide easy access to a wide variety of 5-substituted 6-azauracils.

Acetone sensitized irradiation of 6-azauracil (**1a**) or 1,3-dimethyl-6-azauracil (**1b**) in the presence of vinyl acetates (**2a** or **2b**) produces good yields of the corresponding azetidines (**3a-d**). While these epimeric adducts can be isolated by careful silica gel chromatography, it is more convenient to treat the crude product with water and obtain the dihydro derivatives (**4a-d**) in overall yields of 66-83%. Oxidation of the dihydro derivatives (**4a-d**) affords the respective 5-substituted 6-azauracils.

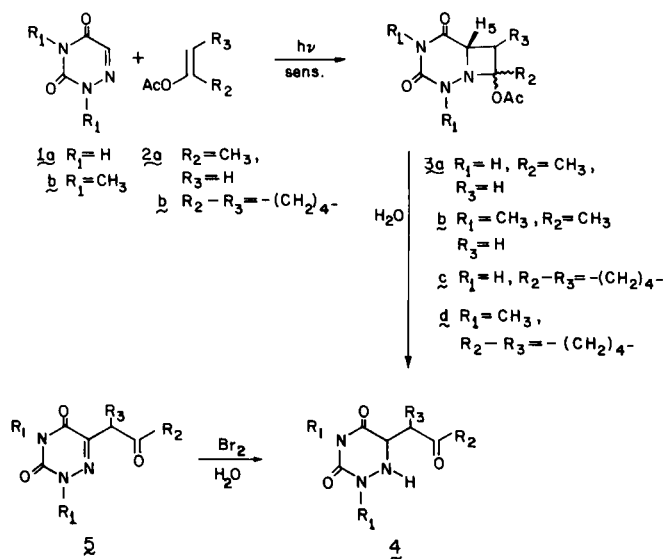
There are two possible orientations for the bicyclic azetidine adducts produced from these reactions, the acetate moiety being either at the 7- or the 8-position of the 1,2,4-triazabicyclo[4.2.0]octa-3,5-dione skeleton. To rigorously establish the orientation for one of these reactions, the major cycloadduct from 1,3-dimethyl-6-azauracil and isopropenyl acetate was isolated by column chromatography (65% yield, m.p. 99-102°). This material, which decomposed rapidly at room temperature liberating acetic acid, showed an nmr spectrum consistent only with the orientation shown: nmr (deuteriochloroform, 60 Mz): 4.28 (doublet of doublets, $J = 6.5, 3.0$ Hz, 1H), 3.18 (s, 3H), 3.13 (s, 3H), 2.8-2.5 (m, 2H), 1.94 (s, 3H), and 1.71 (s, 3H) δ . The appearance of H_5 as a doublet of doublets at 4.28 δ rigorously excludes the product with the acetate and methyl group at position 7 as this would have shown H_5 as a singlet. The azetidines formed in the remaining systems were not rigorously characterized but decomposed with water to yield the corresponding dihydro compounds (**4a-d**). This same orientation is strongly supported for these compounds by their facile hydrolyses under mild conditions and the formation of corresponding dihydro compounds. The structures of the dihydro compounds

TABLE I

Acetone Sensitized Addition of 6-Azauracils to Vinyl Acetates (a,b,c)

6-Azauracil Derivative (moles)	Olefin (moles)	Yield of 4	M.p., °C 4	M.p., °C 5
1a (0.019)	isopropenyl acetate (0.30)	83%	191-193	174-176
1b (0.0071)	isopropenyl acetate (0.018)	70%	90-92	102-103
1b (0.0142)	isopropenyl acetate (0.20)	80%		
1a (0.0088)	cyclohexenyl acetate (0.142)	70%	224-226	239-240
1a (0.0095)	cyclohexenyl acetate (0.021)	70%		
1b (0.0071)	cyclohexenyl acetate (0.107)	66%	112-114	128-130

(a) Corex-filtered light from a 450-watt Hanovia medium-pressure source was utilized for all irradiations; (b) In a typical run the indicated amount of reactants in 70% acetone-water (**1a**) or 10% acetone-acetonitrile (**1b**) was irradiated for 2-4 hours to effect complete reaction; (c) All compounds reported here gave acceptable combustion analyses, $\pm 0.3\%$.



(4a-d) and the 6-azauracils (5a-d) are supported by their combustion analyses, ir, nmr, and mass spectra (5).

The photochemical route described here makes potentially available a wide range of 5-substituted 6-azauracil derivatives. As this sequence offers an extremely mild, high yield procedure for carbon-carbon bond formation, it may be exceedingly useful in functionalizing nucleoside derivatives of 6-azauracil as well as other conjugated imine systems. Studies along these lines are currently in progress and will be the subject of future publications.

REFERENCES

- (1) Camille and Henry Dreyfus Teacher-Scholar (1972-1977).
- (2) For leading references see J. Elis and H. Raskova, *Europ. J. Clin. Pharmacol.*, **4**, 77 (1972); J. Skoda, *Progress in Nucleic Acid Research and Molecular Biology*, **2**, 197 (1963).
- (3) For leading references see J. Gut, *Adv. in Heterocyclic Chemistry*, **1**, 189 (1963).
- (4) J. A. Hyatt and J. S. Swenton, *J. Chem. Soc., Chem. Commun.*, 1144 (1972).
- (5) Complete spectroscopic data will be reported in our full manuscript.